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研究成果の概要（和文）：CO酸化反応とNH₃選択的触媒酸化反応のための高活性Auナノ粒子担持固体触媒の開発に成功した。CO酸化反応では、Au/Nb₂O₅およびAu/Ta₂O₅触媒にて、室温で100%のCO転化率の触媒活性が達成された。NH₃の選択的酸化反応では、温度は、反応温度を室温まで下げると同時に、N₂の選択性を90%以上に上げることに成功した。さらに、この詳細な反応メカニズムを調査することで、より優れた触媒の設計と調製を可能にした。研究成果は、基礎研究と環境関連触媒への応用研究の両側面で、有益な知見を提供した。

研究成果の学術的意義や社会的意義

アンモニアは特有の強い刺激臭を有することで知られ、人間の健康と環境に悪影響を及ぼす。国連の持続可能な開発目標（SDGs）の目標6（きれいな水と衛生）と目標11（持続可能な都市とコミュニティ）の達成に適した都市生活の質を向上させるために、室内の空気浄化の需要がより一層高まっている。我々の研究成果は、NH₃-SCOの反応温度を室温まで下げ、その一方でN₂選択性を90%以上に上げることができた。これは科学的に重要であるだけでなく、低エネルギー消費で安全性の高い空気浄化システム開発に技術革新をもたらすことができる。これは、将来的に健全な社会システムを構築する上で大いに貢献できる。

研究成果の概要（英文）：Highly active Au/solid catalysts were successfully developed. 100% catalytic activity for CO oxidation at room temperature was achieved over Au/Nb₂O₅ and Au/Ta₂O₅ catalysts. Reaction temperature for selective catalytic oxidation of NH₃ was successfully decreased to room temperature and meanwhile the N₂ selectivity was increased to more than 90% over Au/Nb₂O₅. More importantly, the detailed reaction mechanisms were investigated and guided the design and synthesis of more excellent catalysts. The research results offer great scopes for both basic science and applications related to air quality improvement.

研究分野：Environmental Catalysis

キーワード：air purification gold catalyst solid acid high catalytic activity NH₃ oxidation high catalytic activity high N₂ selectivity

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1. 研究開始当初の背景

Ammonia is a well-known odor that has a characteristic pungent smell. It has adverse effects on human health and environment. Demand of indoor air purification is increasing in order to improve quality of life in cities which will also be suited for achieving Goal 6 (Clean Water and Sanitation) and Goal 11 (Sustainable Cities and Communities) in Sustainable Development Goals (SDGs) of United Nations.

Selective catalytic oxidation (SCO) of NH_3 to N_2 and H_2O is an ideal technology for its removal. However, the reported catalysts either need high reaction temperatures ($>250\text{ }^\circ\text{C}$) or present low selectivities to N_2 . Decreasing the reaction temperature of NH_3 -SCO to room temperature and meanwhile increasing the N_2 selectivity to more than 90% are scientifically attractive and challenging, which will bring technological innovation in air purification systems.

2. 研究の目的

Developing highly efficient catalysts that exhibit high catalytic activity for NH_3 -SCO at room temperature and high N_2 selectivity at the full conversion of NH_3 ($>90\%$) is the research target which will offer great scopes for both basic science and applications related to air quality improvement.

3. 研究の方法

According to the preliminary results of the PI, acidic metal oxide supported Au catalysts present high N_2 selectivity towards NH_3 -SCO. Besides, catalytic activity of Au catalysts is highly dependent on the size of Au nanoparticles. Designing and thus synthesizing Au/solid acid catalysts with high Brønsted acid sites and small Au size is the key to achieving the research target.

4. 研究成果

4.1 Development of preparation method for highly active solid acid supported Au catalysts.

Highly active Au/solid acid catalyst contains two parts: the solid acid support with high acidity and redox properties, and the Au nanoparticles with small size. It is known that solid acid properties and redox properties differ depending on the crystal structure. As the typical solid acid, Nb_2O_5 and Ta_2O_5 have various crystal structures and their crystalline phase transformed depend on the heat treatment. Thus, we have successfully synthesized various crystal phase Nb_2O_5 and Ta_2O_5 (as shown in **Figure 1**). The influence of crystal phases of Nb_2O_5 and Ta_2O_5 on the preparation of Au catalysts and their catalytic activities for CO oxidation have also been investigated (*J. Jpn. Pet. Inst.*, 2022, 65(2): 58-66).

The perimeter interface between Au and metal oxide has been reported to be the active site for the activation of the oxygen. Thus, the size of Au nanoparticles is an important factor for the high catalytic activity. It is technical difficult to prepare Au/solid acid catalyst by the conventional deposition-precipitation method due to the charge repels between the negatively charged gold precursor $[\text{Au}(\text{OH})_4]^-/[\text{AuCl}(\text{OH})_3]^-$ and the negatively charged surface of solid acid. Therefore, in order to develop highly active Au catalysts that supported on Nb_2O_5 and Ta_2O_5 with different crystal phases, sol immobilization method was used and further developed. As shown in **Figure 2**, 1-dodecanethiol protected gold colloid ($d = 1.8\text{ nm}$) and 2-phenylethanethiol protected gold colloid ($d = 1.2\text{ nm}$) were synthesized as the gold precursor and were successfully deposited on the high-dimensional Nb_2O_5 with the average size of 2.8 nm and 2.1 nm. The effect of the thiolate ligand on the preparation of Au/ Nb_2O_5 was investigated and a small amount of sulfur species still remained on the catalyst after the calcination at $300\text{ }^\circ\text{C}$, which played an important role for preventing the aggregation of the gold nanoparticles to give ca. 2.8 nm in a mean diameter (*J. Catal.*, 2020, 389: 9-18). Furthermore, in order to avoid the use of toxic organic solvents and simplify the preparation method, the water/ethanol soluble poly

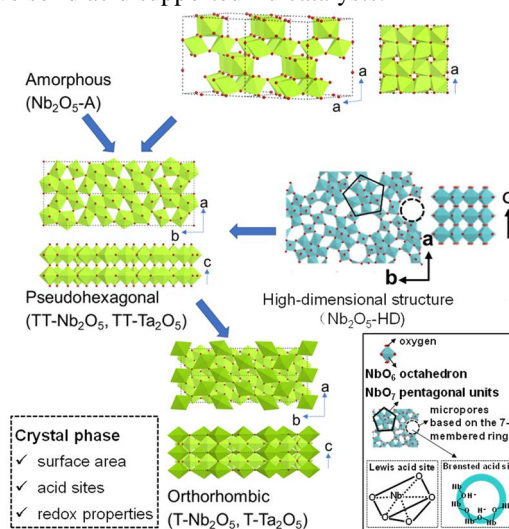


Figure 1. Crystal models of Nb_2O_5 and Ta_2O_5 (Arrows indicate the heat treatment process).

(N-vinylpyrrolidone) (PVP) protected Au-PVP colloid was synthesized and the prepared Au(-PVP)/Nb₂O₅ showed the average Au size of 6.4 nm. Thus, the solid acid—Nb₂O₅ and Ta₂O₅ with various kinds of crystal phases as well as the improved preparation method of gold catalyst were developed and highly active Au/Nb₂O₅ and Au-Ta₂O₅ were successfully obtained.

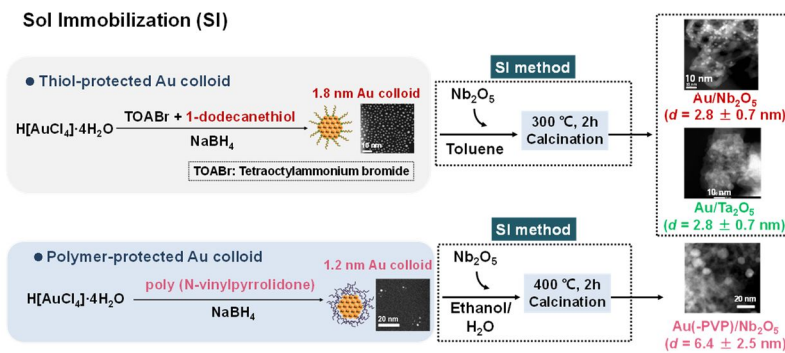


Figure 2. Sol immobilization method by using different ligand-protected Au colloids as the Au precursor.

CO oxidation was selected as the model reaction to evaluate the catalytic activity of the prepared Au/Nb₂O₅ and Au-Ta₂O₅ catalysts. As shown in **Figure 3(a)**, Au/Nb₂O₅ and Au-Ta₂O₅ that prepared by using thiolate protected gold colloid as the gold precursor showed extremely high catalytic activity for CO oxidation. The reasons might be the small sizes of Au nanoparticles which were 2.8 nm, and the high-dimensional Nb₂O₅ (Nb₂O₅-HD) and pseudohexagonal Ta₂O₅ (TT-Ta₂O₅) would provide more oxygen defect sites. The detailed reaction mechanism (**Figure 3(b)**) was studied over Au-Ta₂O₅ with different crystal phases via *in situ* DRIFT measurements (*ACS Catal.*, 2020, 10(16): 9328-9335). Besides, although the Au size of Au(-PVP)/Nb₂O₅ was a little bit large (average size of 6.4 nm) compared to Au/Nb₂O₅ that prepared from thiolate ligand (average size of 2.8 nm), the catalytic activities for both gas-phase CO oxidation and liquid-phase furfural oxidation were also reasonably good (*Catal. Today*, 2022, doi: <https://doi.org/10.1016/j.cattod.2022.03.034>).

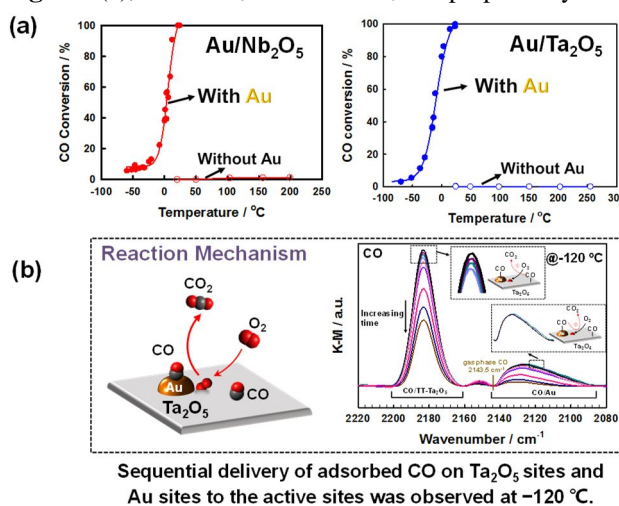
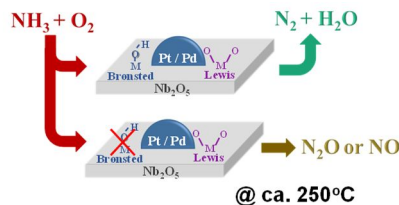


Figure 3. High catalytic activity of Au/Nb₂O₅ and Au-Ta₂O₅ for CO oxidation.

4.2 Highly active catalysts developed for selective catalytic oxidation of NH₃ (NH₃-SCO).

Developing efficient catalysts that can work at room temperature with high N₂ selectivity in ambient air is a daunting challenge, which would offer great scopes for applications related to air quality improvement and control of odors. However, the reported catalysts either needed high reaction temperature (T>250 °C) or showed low selectivity to N₂ with the formation of NO_x as the byproduct. We have found that the Au/Nb₂O₅ catalyst that using high-dimensional structure Nb₂O₅ (Nb₂O₅-HD) as the support with the Au size of 2.8 nm showed room-temperature catalytic activity for NH₃-SCO and high N₂ selectivity at the 100% conversion of NH₃ at 250 °C (>95%). The high-dimensional structure Nb₂O₅ (Nb₂O₅-HD) consists of NbO₆ octahedra, NbO₇, and micropores based on the 7-membered ring in its structure with corner-sharing in the *c*-direction, which notably has the feature of large surface area that exceeds 200 m²/g and high acid amount of Brønsted acid sites, making it to be a good solid acid support (*ACS Catal.*, 2019, 9(3): 1753-1756). In addition, the detailed reaction mechanism revealed that the Brønsted acid sites played an essential role for N₂ selectivity and small Au size was responsible for the room-temperature activity. According to the finding that Brønsted acid sites were important for N₂ selectivity of NH₃-SCO, Nb₂O₅-HD was selected as the support for the preparation of Pt and Pd catalysts. In the previous report, the Pt and Pd catalysts suffered the problem of low selectivity to N₂ in NH₃-SCO, but Pt/Nb₂O₅ and Pd/Nb₂O₅ showed 98% and 100% selectivity to N₂ at the 100% NH₃ conversion, respectively (*J. Catal.*, 2020, 389, 366-374). The N₂ selectivities greatly decreased after utilizing NaOH to block Brønsted acid sites of Pt/Nb₂O₅ and Pd/Nb₂O₅



Catalyst	NH ₃ conv.	N ₂ select.
Pt/Nb ₂ O ₅	100%	98%
Pd/Nb ₂ O ₅	100%	100%

Using Brønsted acid sites of Nb₂O₅ greatly improved the N₂ selectivity of Pt and Pd catalysts (>98%)

Figure 4. NH₃-SCO over Pt/Nb₂O₅ and Pd/Nb₂O₅.

catalysts, suggesting the important roles of Brønsted acid sites. Furthermore, ZHM20 supported Au-Pt alloy nanoparticles catalysts further proved this conclusion (*J. Catal.*, 2021, 402: 101-113).

In addition to the research results of that the Brønsted acid sites are extremely important for the N_2 selectivity in NH_3 -SCO. We investigated the size/structure effect of Ag particles on the NH_3 -SCO. The results showed that Ag/MnO_2 with Ag particles of 2.4 nm diameter provided the highest density of Ag particles over rod-shaped MnO_2 during the reaction, which would be favorable for the formation of adsorbed NO and NH_2NO intermediates to form N_2 and H_2O . Therefore, the Ag size/structure plays an essential role in controlling the N_2 selectivity and the type of byproduct. This work provides a new angle in designing highly active catalysts for NH_3 -SCO with high selectivity to N_2 (*ACS Catal.*, 2021, 11(14): 8576-8584). Based on this finding, Ag/Al_2O_3 with high catalytic activity and high N_2 selectivity for NH_3 -SCO was developed and highly dispersed Ag species supported on Al_2O_3 was found to present > 99% N_2 selectivity at the temperature less than 200 °C. More importantly, this work not only provides a convenient and effective approach to obtain excellent N_2 selectivity in NH_3 -SCO but also affords a systematic insight into the reaction pathways over Ag/Al_2O_3 catalysts (*ACS Catal.*, 2022, 12: 6108-6118).

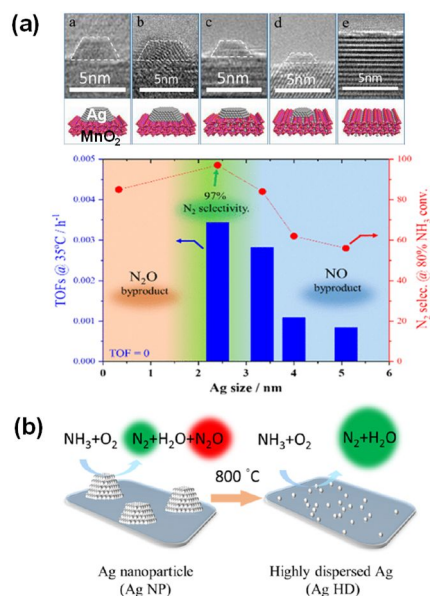


Figure 5. NH_3 -SCO over Ag catalysts.

5. 主な発表論文等

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4. 発表年 2019年

〔図書〕 計0件

〔出願〕 計1件

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〔取得〕 計0件

〔その他〕

6. 研究組織

氏名 (ローマ字氏名) (研究者番号)	所属研究機関・部局・職 (機関番号)	備考

7. 科研費を使用して開催した国際研究集会

〔国際研究集会〕 計0件

8. 本研究に関連して実施した国際共同研究の実施状況

共同研究相手国	相手方研究機関
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