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研究課題名(和文) Topological Defect Engineering on Carbon Electrode for Advanced Batteries

研究課題名(英文) Topological Defect Engineering on Carbon Electrode for Advanced Batteries

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研究成果の概要(和文)：Li-O₂電池の正極におけるトポロジカル欠陥の触媒能を研究した。CVD法により、トポロジカル欠陥が豊富かつエッジサイトのないグラフェンメソスポンジ(GMS)を作製した。GMS正極は従来と比較し、高容量かつ優れたサイクル特性をもつ。電気化学質量分析と理論計算から、非晶質Li₂O₂に対する特異的な触媒能が3.6 V(vs. Li/Li+)の低電荷プラトーに寄与していることがわかる。加えて、これらの特性からGMSは固体触媒の担持に優れるといえる。また、自立型GMSシートは793 Wh kg⁻¹の高いエネルギー密度と0.4 mA cm⁻²の実用範囲での優れたサイクル特性(>260サイクル)を示す。

研究成果の学術的意義や社会的意義

GMSの合成と活用の成功により、これまで報告されていなかったエッジサイトの影響がない、実際のLi-O₂電池のトポロジカル欠陥の触媒能についての知見が得られた。我々はRuを担持したGMSによってトポロジカル欠陥と固体触媒の逐次反応作用を提案した。GMSシートが示す優れた電池性能から、炭素正極の構造を適切に設計することでLi-O₂電池の主要な問題のほとんどを解消できると示された。Li-O₂電池の問題点であったエネルギー密度とサイクル特性は適切に設計された炭素電極により解消された。この結果から、電気自動車用の高いエネルギー密度をもったLi-O₂電池の実用化が促進されると期待される。

研究成果の概要(英文)：In this research, we investigated the catalytic effect of topological defects in carbon cathodes on Li-O₂ batteries for the first time. In brief, topological-defect-rich and edge-site-free graphene mesosponge (GMS) was synthesized by chemical vapor deposition. Compared with conventional carbons, the GMS cathode shows larger capacity and better cycling stability in Li-O₂ batteries. In situ isotopic electrochemical mass spectrometry and theoretical calculations reveal the unique catalysis of topological defects in the formation of easily-decomposable amorphous Li₂O₂, contributing to a low charge plateau around 3.6 V (vs. Li/Li+). In addition, GMS with abundant topological defects and a large surface area proved to be a good substrate for uniform loading of solid catalyst. Finally, the free-standing GMS-sheet with hierarchical pores shows an ultra-high energy density of 793 Wh kg⁻¹ and excellent cycling stability (> 260 cycles) at a practically high current density of 0.4 mA cm⁻².

研究分野：エネルギー関連化学

キーワード：Topological Defect Carbon Materials Graphene Mesosponge Lithium-Oxygen Battery

1. 研究開始当初の背景

Global warming caused by the emission of greenhouse gases has become one of the thorniest problems threatening the survival of mankind. New energy conversion/storage devices are urgently needed to reduce fossil fuel's depletion. Nonaqueous lithium-oxygen batteries with high theoretical energy density and high discharge potential are becoming an attractive candidate for next-generation energy devices. (Energy Environ. Sci., **2019**, 12, 887-922.) A typical nonaqueous Li-O₂ battery consists of a Li metal anode, an organic electrolyte, and a gas-breathing porous carbon cathode. Edge sites in conventional carbon are believed to redistribute the local electron density and increase catalytic activity. (Adv. Mater., **2016**, 28, 6845-6851.) In addition, intrinsic topological defects of carbon materials, such as pentagonal and heptagonal carbon rings, lead to local rehybridization of π -orbitals and create potentially active sites that are beneficial for ORR in the aqueous system. (ACS Nano, **2011**, 5, 26-41.) However, in a nonaqueous Li-O₂ batteries operating at high voltage, edge sites inevitably lead to severe carbon corrosion. (J. Am. Chem. Soc., **2013**, 135, 494-500.) Therefore, to reduce the discharge-charge overpotential, non-corrosive carbon materials with many catalytic sites but few edge sites are urgently needed to catalyze the formation and decomposition of insulated discharge products in nonaqueous Li-O₂ batteries. Prior to this research, no one has attempted to use only topological defects without edge sites as active sites in Li-O₂ batteries.

2. 研究の目的

The proposed research aims to explore the catalysis of topological defects in Li-O₂ batteries and to improve battery performance by using a corrosion-free carbon cathode with topological defects.

3. 研究の方法

This research was conducted under the following five aspects: Material synthesis, topological defect characterization, electrocatalytic mechanism, theoretical calculation, and electrochemical performance.

In FY 2022, I focused on the synthesis of GMS and the characterization of topological defects. In terms of material synthesis, chemical vapor deposition (CVD) and high-temperature treatment (1800°C) were used to synthesize 3-D single-layer GMS using Al₂O₃

nanoparticles as a hard template. (*Adv. Funct. Mater.*, **2016**, 26, 6418–6427.) In addition, free-standing GMS-sheet were also synthesized by CVD. For defect characterization, advanced temperature-programmed desorption (TPD) at up to 1800 °C was used to quantify the edge-sites in GMS. N₂ adsorption/desorption tests were used to determine the BET (Brunauer-Emmett-Teller) surface area of GMS. The intensity ratios of the D-band and G-band in Raman spectra were used to quantify the defect of carbons. Atomic-resolution transmission electron microscopy (TEM) and Scanning Electron Microscope (SEM) were also used to identify the microscopy of topological defects in GMS.

In FY2023, I focused on the catalysis of topological defects and the impact on Li-O₂ battery performance. Experimental in-situ and quantitative characterizations as well as theoretical calculations were performed to investigate the catalysis of topological defects. Differential electrochemical mass spectrometry (DEMS) provides real-time information on gas consumption and generation during the discharge-charge process. Density functional theory (DFT) help to understand the reaction processes at the topological defects. Finally, the performance of Li-O₂ batteries based on GMS, GMS-Ru, and optimized free-standing GMS-sheet cathodes was tested with our home-made device to investigate the influence of topological defects on the performance of Li-O₂ batteries.

4. 研究成果

In FY 2022, GMS was synthesized by CVD at using a CH₄ as carbon source and spherical Al₂O₃ nanoparticles as hard template (*Adv. Sci.*, **2023**, 10, 2300268.). Advanced TPD was used to precisely quantify the amount of H and O edge sites that are desorbed as H₂, CO, CO₂, and H₂O gases as the sample is heated to 1800 °C (inset of Figure 1a). The GMS treated at 1800 °C showed only traces of gas evolution, 0.09 mmol g⁻¹, which was the lowest value among the reference carbon materials (Figure 1a), indicating the edge-site-free property of GMS. As shown in Figure 1b, the intensity ratio of the D and G bands (ID/IG ratio) of GMS is 1.45, which is higher than that of the reference carbon (0.87-1.30). This result demonstrates the topological-defect-rich property of GMS, which is supported by the observation of 5,7-membered rings by TEM (Figure 1c). Edge-site-free and topological-defect-rich GMS is promising as a carbon cathode in Li-O₂ batteries. In addition, GMS with Ru nanoparticles as solid catalysts, GMS-Ru, were synthesized by chemical refluxing. Free-standing GMS-sheets without binders were also synthesized using Al₂O₃-sheets as templates. The textural properties of GMS-Ru and GMS-sheets were systematically characterized.

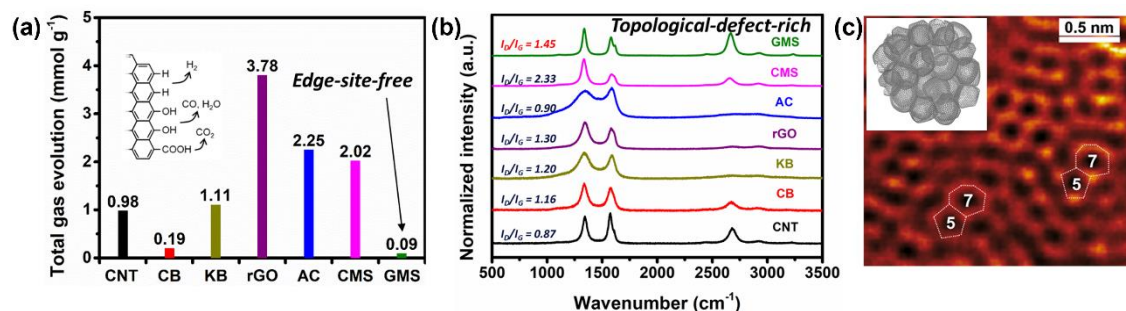


Figure 1 Characterizations of GMS and reference carbons. (a) TPD, (b) Raman spectra, and (c) TEM image of GMS.

In FY 2023, galvanostatic discharge-charge tests were performed with a limited capacity of 0.25 mAh and a current of 0.1 mA to evaluate the cycle stability of the carbon cathodes. The edge-site-free and topological-defect-rich GMS-CP cathode exhibited the longest cycle life. The cycle numbers of the Li-O₂ batteries are plotted against TPD total gas evolution and S_{BET} , as shown in Figure 2a. This implies that GMS is a unique and promising carbon cathode for use in Li-O₂ batteries, showing superior stability due to its edge-site-free property and good catalytic performance due to its topological-defect-rich nature. Based on our DFT calculation, the adsorption affinity of LiO₂ for topological defects was stronger than that on the hexagonal rings. Owing to the different adsorption affinity of LiO₂ on hexagonal rings and topological defects, two different Li₂O₂ formation mechanisms occur in GMS-based Li-O₂ batteries, leading to the formation of typical Li₂O₂ toroids and easily-decomposable Li₂O₂ nanosheets (Figure 2b). Moreover, when we used GMS as a carbon substrate for Ru loading (J. Phys. Chem. C **2023**, 127, 6239–6247), GMS-Ru exhibited unique sequential catalysis with a low charge plateau owing to topological defect and a reduced second charge plateau owing to Ru catalyst. In our recent work (Adv. Energy Mater. **2024**, 14, 2303055), free-standing GMS-sheet shows an optimal energy density of 793 Wh kg⁻¹ and long-cycle stability (> 260 cycles) at a current density of 0.4 mA cm⁻². All the above results prove that GMS with topological defects is one of the best carbon cathodes for Li-O₂ batteries.

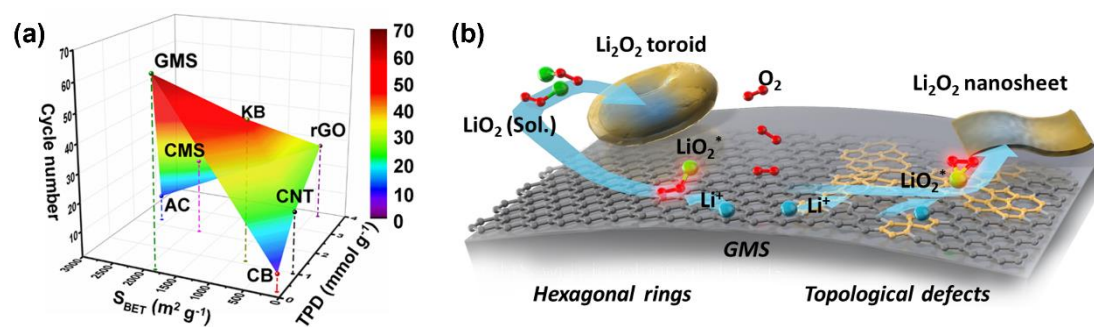


Figure 2 (a) Relationship between the physicochemical properties of carbon materials and the cycle performance of Li-O₂ batteries, (b) Schematic illustrations of discharge mechanism for GMS cathodes.

5. 主な発表論文等

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| 2. 論文標題 Hierarchically Porous and Minimally Stacked Graphene Cathodes for High Performance Lithium-Oxygen Batteries | 5. 発行年 2024年 |
| 3. 雑誌名 Advanced Energy Materials | 6. 最初と最後の頁 2303055 |
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〔出願〕 計2件

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〔取得〕 計0件

〔その他〕

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6. 研究組織

| 氏名 (ローマ字氏名) (研究者番号) | 所属研究機関・部局・職 (機関番号) | 備考 |
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7. 科研費を使用して開催した国際研究集会

〔国際研究集会〕 計0件

8. 本研究に関連して実施した国際共同研究の実施状況

| 共同研究相手国 | 相手方研究機関 |
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