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研究課題名(和文) BNCTに用いるホウ素化合物のin vivo濃度定量法の確立及び薬物動態の把握

研究課題名(英文) Development of an analytical method for in vivo individual concentration determination and the confirmation of pharmacokinetics properties for the two BNCT boron compounds

研究代表者

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研究成果の概要(和文)：本研究はBNCTの臨床に用いられているホウ素化合物のBSH及びBPAをラットに投与し、血漿試料中におけるBSH及びBPAのin vivoホウ素化合物別の濃度測定法を検証した。単剤投与または合剤投与の実験結果より、ラット血漿試料中にBSH及びBPAのホウ素濃度の変化傾向について確認した。また、BSH及びBPAの投与量を変化させることにより、2つのホウ素化合物の薬物動態特性が明らかになった。ラット血漿中にtotalの10B濃度はBSHとBPAの投与割合に関連することが確認された。

研究成果の学術的意義や社会的意義

2剤が併用されるBNCTの臨床において、実験データが極めて少ない状況の中、本研究はホウ素化合物を標的にして、異なる2種類のホウ素化合物を弁別した濃度測定法の確立、そして生体内における各々の薬物動態特性を把握した。これらの研究成果は2剤が併用されるBNCTの臨床治療効果の事後検証の精度向上に期待ができ、学術的な意義が大きい。

研究成果の概要(英文)：In order to develop an effective analytical method to distinguish the 10B concentration from the different boron compounds of BSH and BPA applied in BNCT clinical trials, BSH and BPA were administered to Wistar rats in this research. According to our previous report, the tendency of BSH and BPA in rat plasma is investigated for the single or combined administration used by a LC/MS analytical method. Furthermore, the dose of BSH and BPA were adjusted in different ratio to examine the pharmacokinetic properties of the two boron compounds. It is confirmed that total 10B concentration existed in rat plasma is related to the dose ratio of BSH and BPA.

研究分野：分析化学

キーワード：BNCT BSH BPA LC/MS ホウ素濃度

A>>7D> 7> >7 >7> A>>7> 88  
>>' 268bũ  
BNCT >> 3p+Z (b) i  
10B BSH (Sodium Borocaptate) \ BPA (p-Boronophenyl al anie) b 2 ' 8b(i) 7b BNCT +< \_  
BNCT 87+> 12 10B NA 10B b3z3 8  
@ /b0v(y+w348bfKPKZ3,  
rAc8D1 @MSu0p+K(6Z66  
S BPA c8r ABSBj/ b+K4E  
Z3t1 E BSH ACB @ BPA (p 10B cN? 5  
Su+KES (ò) BMCsbGpGbs  
u BNCT +% ' 2E1Q0+K 0cv3sfi 2  
K .cGMm 2GZGbc(b) (ò  
> 10B> b 200 Gr[ (ò) 2KZ 1) 1  
(ò) ICP-AES rSc 1) A5 (ò) ICP-MS @ M+ 7  
KGB 2c 2 cGf I BNCT bc (ò) K  
Z 5 MG\ @ A8 Su total b 10B KZb@ #yb ' 6  
p^ W8 8dAp' 2a & INMRC> 87\$7T0  
+^ 382' 2at bdc' 2)g BNCT b  
+< 2 K (ò) Hb 2b' 26SW  
(ò) (y| bv3s1 u)) 10B 6W  
no(8) H. Kumada, 2015> BNCT MSu  
H\* /E(ò) i DBC +9Z 250-500mg/kg 2 6  
BPA 6KZGr b M BSH bCZ 2 cGf Q  
B/Gc4) Z 8SuZhb=KZ  
7 6M9euZ^  
Gv6 & K. Yokoyama, 2006>

0>' 2b% \$  
' 2 C (H25 H27) \_ FLKS18 937A5  
(ò) (LC/MS) Z 2 (b(7fg 2&gKS %' 2c G  
b5(ò) BNCT b +< \_ 4f 0Su p 2+ 6pb(ò)  
2&gM BSH Ig BPA b. 6G% 6

1>' 2b2  
' 22KZ q p zS/ 0bS#0  
b40/WS r p Z [ GvKS/ 0Z 5 (ò 2b 6  
/0q 1b4KS s BSH Ig BPA QRrSc c KS 2  
/ c 0 BSH Ig BPA b)i i & KS tV/E 2  
(ò) MSu\_ total b 10B QR BSH \  
BPA bZ 1K Z )sOS 3S

& 1> 5) zS/ 0bS# 0 bqi 0f  
□ □ □ □ BSH> BPA> b^ a = Pfi KZ p 42  
p 5 )z8BK 0 /WZ m^ 7NKS / 7V  
(7WZ#X b(#8V% \$9ps\_ // \$ 38 1>/WS OKZ / 0q 6 7ub 4#1 gb  
& -20% 2 X b 2N K 2 46° 5)zR 6 7ub 6q  
10K Sbu LC/MS ) B& & n=6> /WS 0q 8Bm8p@ / 0q 2 46°  
(0s8j mSuf Gb S#b 8m8p3KZ / 0q 2 46°  
& i fSc pKS 5)z8B%\$/%010 Z 7V#M  
8: 2 0 Z 0E 4 K 5)z [ 300KS 5)zb 0 (VZ BSH c  
0.25# g/ml 5.0# g/ml & LLOQ8 0.25# g/ml > BPA c 1.0# g/ml 25.0# g/ml & LLOQ8 1.0  
# g/ml > uZ  
(MSu\_ OC 0q S OC 0q c 0q p p  
9x 3 LL0Q& 5W7H > Z 4 I b 0 QR& BSH+BPA>b^ a=  
P6 BSH-OC b 0.25 0.75 1.50 4.00# g/ml BPA-OC b 1.00  
7.50 15.00 22.50# g/ml 10 KS 5)zb %zö % RE#1=PM  
%R1> Ig OC b )Y & CV Ig RE> #0 b qi 0f KS rS G  
0q b BSH Ig BPA b 3K 0 & n=6> &K S

& 2> 0 0q 1b4 0f

BPA cPM P00Su BNCT b+ $\xi$  BPA  $\gamma$  & Wc BPA-fr $\delta$   
 g'  $\xi$  LC/MS  $\downarrow$  BPA-fr b $\xi$  BPA Ig BPA-fr b = Pfi QR 10K LC/MS  
 (a) = Pfb BPA-fr b $\xi$  BPA Ig BPA-fr b = Pfi QR 10K LC/MS  
 (b)  $\xi$  BPA-fr b $\xi$  & BSH $\gamma$  BPA $\gamma$  bc Z  $\xi$ vKS  $\xi$  /  $\xi$   
 LC/MS  $\downarrow$  BSH Ig BPA & BPA-fr bg'  $\xi$ S $\gamma$  b Q Rb  $\xi$ -  
 &KS

& 3 $\gamma$   $\xi$  BSH Ig BPA b  $\xi$  ) $\gamma$  Ig  $\xi$   
 & Wistar rat  $\xi$  5 4K/5 100g S  $\gamma$  BPA  $\xi$  & 250mg/kg $\gamma$   
 \ BSH  $\xi$  & 100mg/kg $\gamma$  Ig gb c & BPA250mg/kg $\gamma$  BSH100mg/kg $\gamma$  [ 7 $\xi$   
 KZ b 0.5h 2.5h 6.0h H $\downarrow$  G//WS 42(7 $\xi$  5000rpm 10min., 4 $\gamma$   $\gamma$  -  
 WZ  $\xi$ # [ (KS/  $\xi$ (bs P1 $\xi$  S Q R 10KS5 $\gamma$   
 b /  $\xi$   $\xi$  /8  $\xi$ / $\xi$  BSH \ BPA b)i  $\xi$   $\xi$  &  
 1KS  $\gamma$   $\xi$  BSH Ig BPA b  $\xi$  c-  $\xi$ Su BSH Ig BPA  
 b $\xi$ )zb5(V pr $\downarrow$  Ob $\xi$   $\xi$  S $\xi$  KS

& 4 $\gamma$  /  $\xi$  BSH Ig BPA b.  $\xi$  1 $\xi$   
 BSH Ig BPA b.  $\xi$  Su $\downarrow$   $\xi$ B  $\xi$ f P1 $\xi$  KZ & BSH+BPA $\gamma$ cZ  
 total b  $\xi$ B  $\xi$ c $\xi$ /56S- 60 mg/kg  $\xi$  BSH \ BPA b $\xi$   
 1 $\xi$  5  $\xi$  Z [  $\xi$ KS  $\xi$ Z Z b 0.5h Ig 2.5h  $\xi$  Gv  
 KS/  $\xi$  QC  $\xi$  v /WS & n=3 $\gamma$

2 $\gamma$ ' 2B $\gamma$   
 & 1 $\gamma$ z $\xi$ /  $\xi$  S#  $\xi$  ó b $\xi$   $\xi$ f  
 $\xi$   $\xi$  BSH Ig BPA bg $\xi$   $\xi$   $\xi$ @  
 & R $\xi$  0.99 $\gamma$   $\xi$ z b $\xi$  RE $\gamma$   $\xi$ Z 7V $\xi$ #KS  
 $\xi$ fSc $\xi$  /  $\xi$   $\xi$  5( $\xi$ b8 $\xi$ = & LLOQ  $\xi$ % $\xi$   
 s 20 $\xi$  LLOQ  $\xi$ % $\xi$  15 $\xi$  6SK $\xi$  $\xi$   $\xi$ KS  
 - 7V $\xi$ #KS /  $\xi$   $\xi$  BPA b LLOQ & 5W7H  $\gamma$   $\xi$ % $\xi$  & RE $\gamma$   
 b " & 52 $\xi$  c 5( $\xi$ b8 $\xi$ = 6SK $\xi$ G $\xi$ S  
 r  $\xi$ z $\downarrow$  QC  $\xi$   $\xi$  Ig LLOQ b 4  $\xi$  Qrb $\xi$  & RE $\gamma$  Ig  
 ( $\xi$  & CV $\gamma$  'K S  $\xi$  3  $\xi$  LLOQ & 5W7H  $\gamma$   $\xi$  QC b RE I  
 g CV c 30 $\xi$  - A $\wedge$   $\xi$ KS & Table1 $\gamma$  G $\xi$  ) $\gamma$  ( $\xi$  M S  
 u\_ / $\xi$ ( $\xi$ 9 /G Table1  $\xi$ )zb LLOQ  $\xi$  QC b ) $\gamma$

$\xi$ #b/  $\xi$  Mf  $\xi$   $\xi$   
 5)z8Bb % $\xi$ /010 $\xi$  ST  
 $\xi$ bc LLOQ  $\xi$ b\* $\xi$  ( $\xi$ )  
 & M $\xi$   
 s QC  $\xi$  b $\xi$  & n=6 $\gamma$   $\xi$ Z 3  $\xi$   
 [ KS) $\gamma$  c  $\xi$ Nv CV c  $\xi$   
 5 $\xi$   $\xi$  [ 6W  
 S rS $\xi$  & RE $\gamma$  b5) $\xi$ 5( $\xi$ )  
 2b8 $\xi$ = 6SK $\xi$ G $\xi$ KS

前処理手順	冷蔵または凍結融解保存後 除タンパク前処理実施		除タンパク前処理実施後 冷蔵または凍結保存	
検量線	検①	検②	検③	検④
温度条件	4°C	-20°C	4°C	-20°C
CV%(BSH)	1.4	3.0	1.5	1.7
CV%(BPA)	2.3	8.4	3.5	1.8
RE%(BSH)	-14.8	-15.9	-18.9	-33.4
RE%(BPA)	-21.0	57.4	-12.3	8.6

& 2 $\gamma$   $\xi$ b5( $\xi$  b4 $\xi$ f  
 q= Pfb BPA-fr b $\xi$   
 = Pfb BPA-fr b $\xi$   $\xi$  &KS BPA-fr c BPA  $\downarrow$   $\xi$   
 $\xi$  ESI  $\xi$  m/z 207  $\xi$ S rS BPA-fr \ BPA b $\xi$  3Q  
 K SvB Fig.1  $\xi$ KS L  $\xi$ KS BPA-fr \ BPA c  $\xi$ g c 5a  
 $\xi$  op M+K $\xi$ G $\xi$  ? rS $\xi$  0.15ml/min. \ 0.20  
 ml/min.  $\xi$ g \$  $\xi$  1# I 2# I 3# I  $\xi$ QR BPA-fr \ BPA b $\xi$ 3Q  
 KS@ 8Nv  $\xi$  s 0.1min.  $\xi$  M+K $\xi$  G $\xi$ KS

r $\xi$  BPA-fr b $\xi$   
 & BSH $\gamma$  BPA $\gamma$  b cKS /  $\xi$   $\xi$   $\xi$ Z %  $\xi$  (b $\xi$  &KS  
 & Fig.2 $\gamma$  BPA-fr c 5min. S $\xi$  ESI ( $\xi$ ) m/z 207  $\xi$  BSH c 18min. S[  
 $\xi$  ESI (+) m/z 723  $\downarrow$   $\xi$  W $\xi$ ( $\xi$ ) 2 $\xi$ gKS( $\xi$ )  
 $\xi$  c  $\xi$   $\xi$

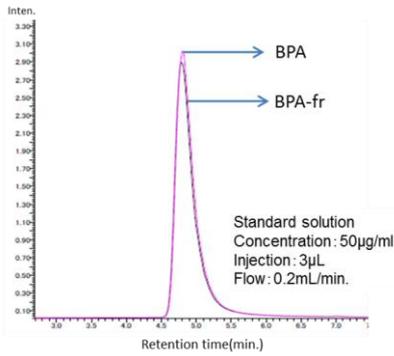


Fig.1 BPA-fr BPA

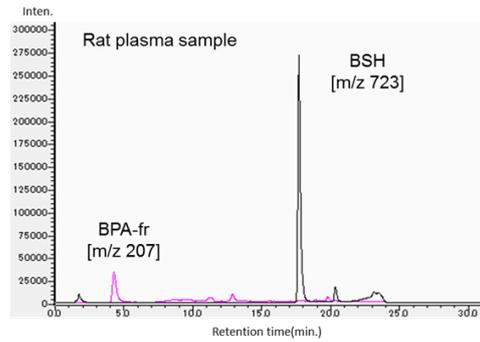


Fig.2 BPA-fr BSH

Table 2.  $^{10}\text{B}$  concentrations in rat blood at different time points after administration of BPA, BSH, or their combination. The data shows that BSH is rapidly eliminated, while BPA remains in the blood for a longer duration. The combination of BPA and BSH shows a faster elimination of BPA compared to BPA alone.

Rat blood collection time points	BPA single administration (250mg/kg)		BSH single administration (100mg/kg)		Co-administration of [BPA+BSH] (BPA250mg/kg+BSH100mg/kg)	
	BPA Conc. [ $\mu\text{g/ml}$ ] (mean $\pm$ SD)	rat	BSH Conc. [ $\mu\text{g/ml}$ ] (mean $\pm$ SD)	rat	BPA Conc. [ $\mu\text{g/ml}$ ] (mean $\pm$ SD)	BSH Conc. [ $\mu\text{g/ml}$ ] (mean $\pm$ SD)
0.5h	155.7 $\pm$ 27.6	rat,5	102.4 $\pm$ 12.9	rat,6	125.9 $\pm$ 34.0	103.1 $\pm$ 20.3
	7.5 $\pm$ 1.3		58.5 $\pm$ 7.4		6.0 $\pm$ 1.6	58.9 $\pm$ 11.6
2.5h	78.3 $\pm$ 5.8	rat,5	9.1 $\pm$ 1.9	rat,6	61.2 $\pm$ 8.3	15.2 $\pm$ 1.9
	3.8 $\pm$ 0.3		5.2 $\pm$ 1.1		2.9 $\pm$ 0.4	8.7 $\pm$ 1.1
6.0h	57.3 $\pm$ 6.2	rat,6	0.3 $\pm$ 0.1	rat,5	39.3 $\pm$ 3.5	0.4 $\pm$ 0.4
	2.7 $\pm$ 0.3		0.2 $\pm$ 0.1		1.9 $\pm$ 0.2	0.2 $\pm$ 0.2

Figure 3 shows the pharmacokinetic profiles of BPA in rat blood. Graph (a) shows BPA alone, and graph (b) shows BPA+BSH co-administration. Both graphs plot BPA concentration ( $\mu\text{g/ml}$ ) against time after injection (hour). The data points are consistent with the values in Table 2.

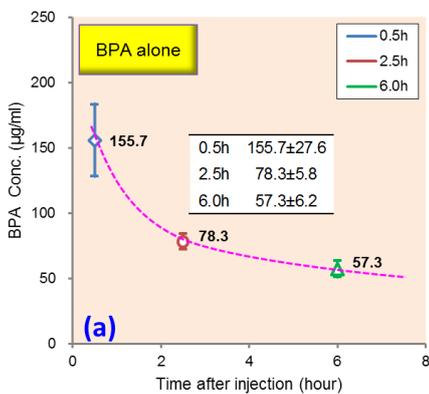
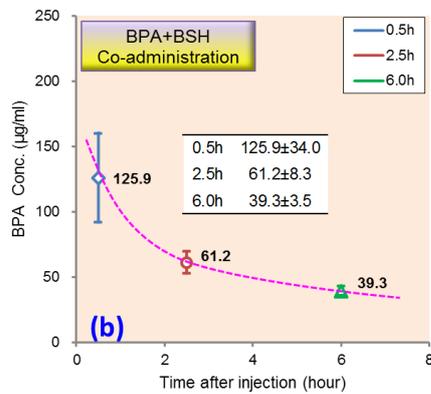


Fig. 3 BPA concentration



BPA concentration

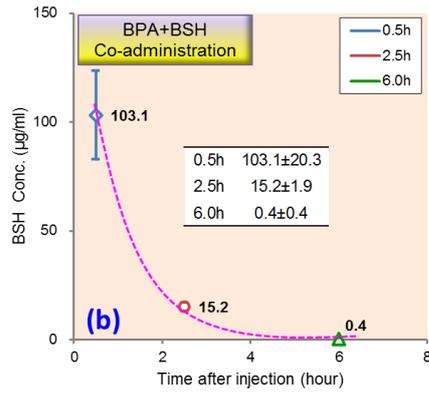
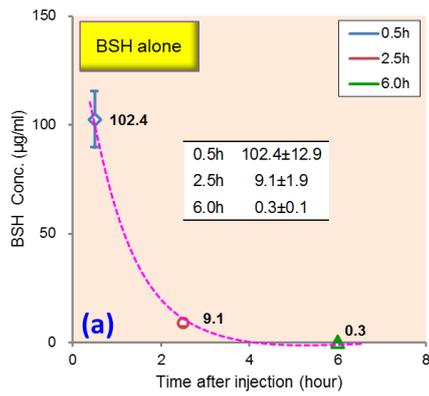


Fig. 4 BSH 浓度

BSH 浓度

BSH 浓度 BPA 浓度  
 BSH+BPA 浓度  
 0.5h 2.5h 6.0h  
 SD Table3\_9KS  
 Fig. 5  
 Fig. 6

Table3. cZ 浓度

<sup>10</sup>B-BSH \ <sup>10</sup>B-BPA 浓度

Injection (BSH+BPA)	After injection 0.5h			After injection 2.5h		
	Mean value±SD (n=6), µg/ml			Mean value±SD (n=6), µg/ml		
	<sup>10</sup> B(BSH)	<sup>10</sup> B(BPA)	Total <sup>10</sup> B	<sup>10</sup> B(BSH)	<sup>10</sup> B(BPA)	Total <sup>10</sup> B
Pattern A [BSH]: 50mg <sup>10</sup> B/kg [BPA]: 10mg <sup>10</sup> B/kg	7.50 ±1.53	4.67 ±1.18	12.17 ±2.08	2.20 ±0.42	3.61 ±0.38	5.81 ±0.32
Pattern B [BSH]: 40mg <sup>10</sup> B/kg [BPA]: 20mg <sup>10</sup> B/kg	7.49 ±1.97	6.55 ±0.79	14.05 ±2.51	1.09 ±0.26	4.31 ±0.45	5.40 ±0.65
Pattern C [BSH]: 30mg <sup>10</sup> B/kg [BPA]: 30mg <sup>10</sup> B/kg	4.66 ±0.89	11.80 ±1.69	16.45 ±2.39	0.94 ±0.20	6.72 ±0.74	7.65 ±0.87
Pattern D [BSH]: 20mg <sup>10</sup> B/kg [BPA]: 40mg <sup>10</sup> B/kg	3.31 ±1.39	24.21 ±7.82	27.53 ±9.11	1.00 ±0.22	9.98 ±0.86	10.97 ±0.88
Pattern E [BSH]: 10mg <sup>10</sup> B/kg [BPA]: 50mg <sup>10</sup> B/kg	2.30 ±0.81	32.53 ±13.37	34.83 ±13.97	0.62 ±0.17	11.74 ±1.65	12.36 ±1.77

total b <sup>10</sup>B  
 c L 浓度  
 v6N BPA  
 b Q X  
 / 浓度 total  
 b <sup>10</sup>B 浓度  
 " c 浓度  
 0.5h 浓度  
 Eb total b <sup>10</sup>B 浓度  
 34.83  
 s 13.97# g/ml >c  
 A & 12.17s 2.08# g/ml > b (U3 p)  
 BPA 浓度 c & 208 1042> mg/kg 8 (V [ )  
 BSH 浓度 & 87.6 17.5> mg/kg b(V 浓度 total b <sup>10</sup>B 浓度  
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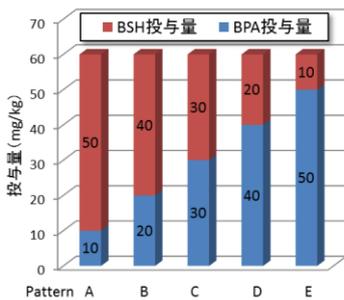


Fig. 5

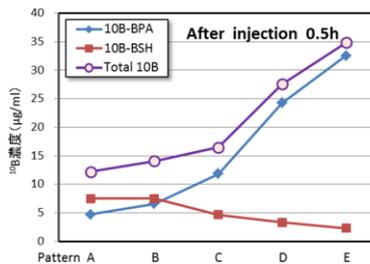
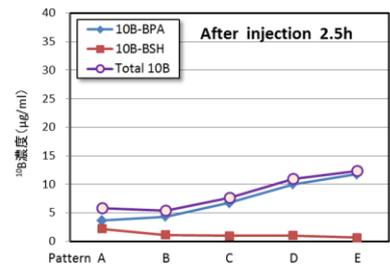


Fig. 6



10B-BSH 10B-BPA Total 10B 浓度

1> 22B% 2BÝ  
/ 2b6

H25> H27> \$ 2

BNCT E/fib  
2

BSH \ BPA 1000

3 > z\$1=e ' ...  
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2 6

q C. Bi, K. Nakai, H. Kumada, Y. Yamaguchi, S. Bamba and T. Morimoto, Verification of boron compounds concentration measurement using LC/MS for combination of boronophenylalanine (BPA) and borocaptate (BSH) in rat plasma, 9<sup>th</sup> Young Researchers' BNCT Meeting, Nov. 13-15, 2017, Kyoto, Japan.

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